[530]

III. Part of a Letter from Mr Geoffrey, F. R. S. to Dr Sloane, concerning the exact quantity of acid Salts contained in acid Spirits.

Paris, June 10. 1699.

Believe M. Monro has written to you, what has been done in our publick meeting the 29th of April, in the Kings Palace called the Louvre, where fince that day we meet weekly, therefore I fend you only a short account of the discourse of Mr Homberg, Chymist of the Academy, his discourse was concerning the exact quantity of Acid Volatile Salt contained in Acid Spirits. And first he gives an account of the exactness required in Chymical Operations, not only in the quality of the matters which are to be mingled, but also in their quantity and doles, and for want of that exactness very often the experiments do not succeed. We have flays he) a very fure way for measuring the quantity of solids, by balances and ordinary weights, but we cannot come to the same precision in Liquors; and yet less, we can't know the precise quantity of the different matters contained in those Liquors, as what is the quantity of Acid Volatile Salt contained in Acid Spirits, for that purpose he has contrived a new Araometer, or a measure of Liquors, of which here is the description.

A fig. 1: is a glass Bottle like a little matracium, of which the neck B C is so small that a drop of Water therein takes up the space of five or six lines, near that neck is a little Capillar tube D about six lines long, and parallel to the neck B C, in the opening B is a little dilated, fashion of a tunnel, for pouring more easily the Liquors in the Bottle, and the little tube D is forgiving a way to to the air contained in that Vessel to go out when the Liquor is poured in B, the point C is a little mark at the same height as the end of the little tube D.

When

[531]

When we fill that Vessel with some liquor for the experiments, we pour the Liquors into the Bottle by the opening B, until it goes out by the little tube D, and if the height of the Liquor is even to the mark C. 'tis well; if it is lower, we must fill more to that point; if it is higher, we must strike softly upon the opening B, till the overplus of the Liquor be even to the point C in the neck of the Bottle. By that means we have always exactly the same volume of Liquor, and we can know how the same volume of the several Liquors weighs more one than another precifely, but as the volume of Liquors is not always the same, and changes according to the alteration of the weather, now colder and then hotter, we must consider the variation of the weather. when we will do the Experiments in feveral times, and when we will compare the weight of the Liquor, which we weigh in Summer time, with the weight of an other, which we have weighed in Winter; for the fame Liquor being more rarefy'd in the hot time, and condens'd in the cold, the same volume of it will be more weighty in cold weather than in warm; for that purpose M. Homberg has given us a Table of the various weights of some more usual Liquors in the coldest time and in the hottest, as it is underneath.

The Areometer full of Mer-\\3\3 gr.		5 gr.
cury, in the Summer time \(xi \) o vi	inWinterxi	OXXXII
Full of Oil of Tartar — i iii viii	i	iii xxxi
Spirit of Urine i o xxxii	i	o xliii
Oil of Vitriol i iii lviii	i	ivi iii
Spirit of Nitre —————— i i xl	i	ilxx
Spirit of Salt i o xxxix		0 xlvii
Aqua fortis — i i xxxviii		ilv
Vinegar. o vii lv	O	vii l x
Spirit of Wine —— o vi xlvii	0	vi lxi
River Water — o vii liii	0	vii lvi i
Distilled Water o vii l	0	vii liiii
This empty Areometer weighs 3i and gr	. xxviii.	It

[532]

It appears by this Table that all Liquors, and likewife the Me cury are able to be condented and rarefy'd in cold and hot weather.

For the quantity of the Volatil Acid Salt contained in the Acid Liquors, M. Homberg declared first what he intends by this Salt, and told us, the Acid Spirits were no other thing but a Salt diffolved by a little water, which the talte shews well enough for an acid, as also his effects, he calls it Volatile, because it is raised by the fire with the Phlegm, and it cannot be but hardly separated from that, and reduced in a dry form: what nevertheless M. Homberg has made in the operation inferted in the Memoirs of the Royal Academy, published the 15th of December 1692. by what operation it appears that the acid spirits are nothing but volatil acid falt and phlegm. The quantity of falt contained in a determined quantity of acid spirit was not yet known. but he has given a way to know it, and also he may fay the quantity of falt contained in whatever acid spirit, only by the weight of volume, compared with the weight of another spirit, of which the quantity of falt contained in it was known, First, for knowing the quantity of volatile acid falt contained in some acid spirits, he has poured upon an ounce of salt of Tartar well dryed the quantity of one acid spirit, as much as the falt of Tartar has been able to take of it, then he evaporated all the infipid humidity or Phlegma out of this falt, and he weighed the matter, the quantity of his weight, above the weight of the falt of Tartar before faturation, is the quantity of acid volatile falt contained in the quantity of acid spirit which has been taken by one ounce of falt of Tartar. Here is the Table of the quantity of acid that has been necessary to the perfect impregnation and fulness of the falt of Tartar, and by the same means the Table of the quantity of acid Volatile Salt, contained in one ounce of feveral acid spirits. For

[533]

For the perfect impregnation of one ounce of falt of Tartar, was poured upon it spirit of Nitre 3i. 3ii. gr. xxxvi. the weight of that salt after the evaporation of the insipid humidity has been encreased to 3iii. gr. x, above one ounce; that increase coming from the acid remained in the salt of Tartar, shews to us that one ounce of spirit of Nitre contains 3ii. gr. xviii. of acid salt.

So for the impregnation of \$\frac{3}{1}\$ of falt of Tartar, has been poured upon it spirit of Salt \$\frac{3}{1}\$ii. \$\frac{3}{2}\$v. the increase after the evaporation has been found \$\frac{3}{1}\$iii. gr. xiv. and therefore one ounce of spirit of Salt contains \$\frac{3}{1}\$i. gr. xv. of acid Salt.

Upon 3i of falt of Tartar has been poured oil of Vitriol 3v. the increase has been found 3iii. gr. v. therefore 3i. of oil of Vitriol contains 3iii. gr. lxv. of acid Salt.

Upon 3i. of salt of Tartar has been poured Aqua fortis 3i. 3ii. gr. xxx. the increase has been found of 3iii. gr. vi. therefore 3i. of Aqua fortis contains 3ii. gr. xxvi. of acid Salt.

Upon şi. of salt of Tartar has been poured distilled Vinegar şxiiii. the increase has been found of ziii. gr. xxxvi. therefore şi. of distilled Vinegar contains gr. xviii. of acid Salt.

It appears by this Table that the quantity of acid Salt for faturating the falt of Tartar is near the fame, though the quantity of acid liquors should be very different; it is only the acid of Vinegar, of which the salt of Tartar retains more than it does of the others, that M. Homberg attributes to the subtilty of the particles of the vegetal acid, which have been very much divided by the alterations in the fermentation of the liquors in the Plants, &c. of the Wine, and also in the distillation, which alterations the mineral acid has not received. The vegetal acid by that subtilty of particles is able to impregnate a greater quantity of Liquor, than

[534]

the same quantity of mineral acid, and by that it is more easily raised up by the fire than the others.

By these observations M. Homberg makes evident the reason of some cases, difficult to be explained without them: as, 'tis well known one ounce of Aqua Regia, compounded with the spirit of Nitre, and the ammoniac Salt, dissolves twice more Gold than one ounce of the spirit of Salt can do. The Chymists attribute that effect to the softness of the points of one acid, and to the hardness of the other: when these observations make evident that the spirit of Nitre contains twice more of acid Salt than the like volume of spirit of Salt, and open in the same time the true cause of this effect.

M. Homberg discours'd also how by the comparison of these two Tables we may know the quantity of acid Salt contained in an acid Spirit; which he doth in the following manner: He takes an acid spirit (as spirit of Nitre) he weighs it by his Areometer, and at the fame time he weighs also distilled water (for the weight of the phlegm contain'd in the acid spirits is like as the weight of the distilled water) then he looks upon the Tables, where he fees that the bulk of spirit of Nitre compared with a like bulk of distilled water, has given a certain quantity of acid Salt for each ounce; and from thence he concludes, that the bulk of other spirit of Nitre, of which the weight is known, compared with the like of water, shall give a determined quantity of acid falt, which will be raifed by the computations of the relations of the weights of those spirits, with the weights of the like bulks of distilled water. by concluding from them and from the known product of acid falt for the unknown product of the same. All this more explained in the account which will be inserted in the Memoirs of the Academy.

IV. Part

Philos : Transact N: 262.

